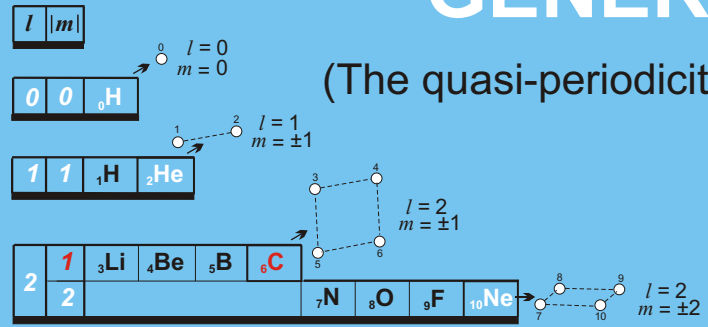


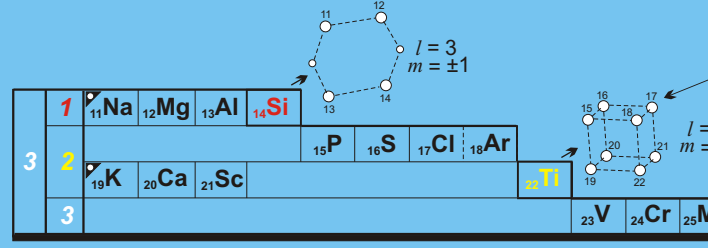
GENERALIZED TABLE OF THE ELEMENTS

(The quasi-periodicity as a result of quasi-similarity of nodal structure of external atomic shells)

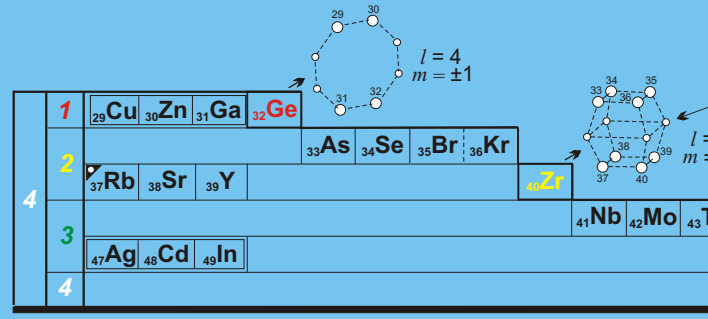


The simplest solutions of the wave probabilistic equation in the spherical polar coordinates. Presented in the form of the spatial distribution of *potential* extremes-nodes (discrete elements of the **shell nucleon structure** of atoms) and in the traditional form of periodic table of the elements [1-3]: $\psi_{l,m}(r, \theta, \phi) = C R_l(r) Y_{l,m}(\theta, \phi) \cos m\phi$ where C is the constant factor, $r = kr$ is the radius of characteristic shells, θ and ϕ are polar and azimuth angles, respectively.

$$\frac{1}{c^2} \frac{\partial^2}{\partial t^2}$$



Numbers 1, 2, 3, ..., 110 are the ordinal numbers of the **principal polar-azimuth nodes** coinciding with the atomic numbers of elements Z.



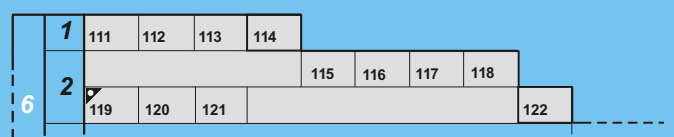
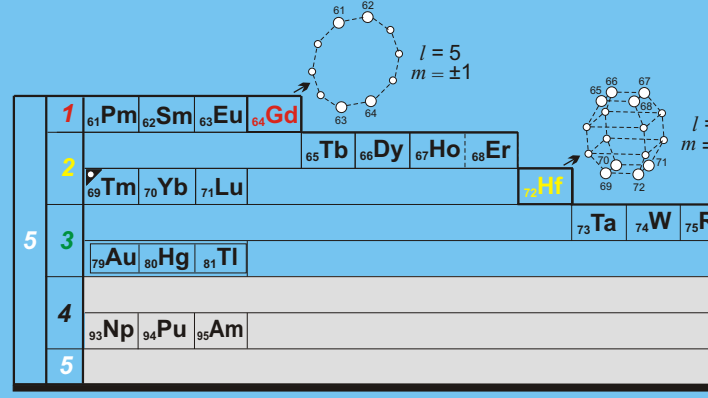
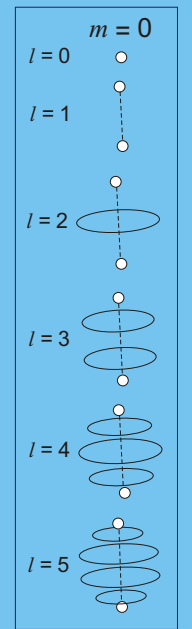
The **collateral polar-azimuth nodes**

Elements with the completely filled outer nucleonic shells

² He				
⁶ C	¹⁰ Ne			
¹⁴ Si	²² Ti	²⁸ Ni		
³² Ge	⁴⁰ Zr	⁵² Te	⁶⁰ Nd	
⁶⁴ Gd	⁷² Hf	⁸⁴ Po	¹⁰⁰ Fm	¹¹⁰ Ch

is the designation of unstable elements

Polar nodes and rings



[1] Alternative Picture of the World, V. 1-3, (1996); [2] Foundations of Physics, (1998); [3] Atomic Structure of Matter-Space, (2001); Geo. S., Bydgoszcz